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Molecular membrane separation: plants inspire new technologies

Authors for correspondence:

Caitlin Byrt

Email: caitlin.byrt@anu.edu.au

Samantha McGaughey

Email: samantha.mcgaughey@anu.edu.au

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Annamaria De Rosa* , **Samantha McGaughey*** , **Isobel Magrath**  and **Caitlin Byrt** 

Division of Plant Science, Research School of Biology, Australian National University, 2601, ACT, Acton, Australia

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Summary

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Plants draw up their surrounding soil solution to gain water and nutrients required for growth, development and reproduction. Obtaining adequate water and nutrients involves taking up both desired and undesired elements from the soil solution and separating resources from waste. Desirable and undesirable elements in the soil solution can share similar chemical properties, such as size and charge. Plants use membrane separation mechanisms to distinguish between different molecules that have similar chemical properties. Membrane separation enables distribution or retention of resources and efflux or compartmentation of waste. Plants use specialised membrane separation mechanisms to adapt to challenging soil solution compositions and distinguish between resources and waste. Coordination and regulation of these mechanisms between different tissues, cell types and subcellular membranes supports plant nutrition, environmental stress tolerance and energy management. This review considers membrane separation mechanisms in plants that contribute to specialised separation processes and highlights mechanisms of interest for engineering plants with enhanced performance in challenging conditions and for inspiring the development of novel industrial membrane separation technologies. Knowledge gained from studying plant membrane separation mechanisms can be applied to developing precision separation technologies. Separation technologies are needed for harvesting resources from industrial wastes and transitioning to a circular green economy.

*These authors contributed equally to this work.

I. Introduction

During the evolution of plants, there was selective pressure to achieve energy efficient precision in membrane separation control to maintain the cellular and organelle homeostasis required for essential biological processes (Hwang *et al.*, 2016). Heterogeneity in the chemical composition of soil across the globe meant that plants were challenged to adapt to different soil solution compositions in different geographical areas (van der Ent *et al.*, 2013). Soil solution compositions with excess amounts of heavy metals, minerals and salts can present a challenge to plant growth by limiting the uptake of essential nutrients, and causing toxicity and oxidative stress (Ali *et al.*, 2019).

Plant adaptation to challenging soil solution compositions can include strategies such as avoidance via root architecture changes; biomodification or detoxification; adopting symbiotic relationships with soil microbes; and enhanced uptake followed by sequestration of undesirable molecules in membrane bound compartments such as the vacuole (Khare *et al.*, 2017). There have been many studies of adaptation mechanisms where transcriptomic, proteomic, metabolomic and ionic approaches were deployed to investigate the genetic and physiological basis of adaptation (Baxter *et al.*, 2012; Singh *et al.*, 2016; Stein *et al.*, 2017). These types of studies revealed that some adaptation strategies involve uptake and sequestration of undesirable elements, and these strategies require specialised membrane separation mechanisms (Yang *et al.*, 2005; Verbruggen *et al.*, 2009). Plant membrane separation processes enable undesirable elements to be distinguished from desirable elements and create the potential for undesirable elements to be effluxed from the plant or compartmentalised in the vacuole or in specific cells (Conde *et al.*, 2011; Pasricha *et al.*, 2021). These processes are important for the survival of life on Earth, and life has evolved a suite of transport proteins that enable this selective separation across different membranes to maintain cellular and organelle homeostasis. For example, of the *c.* 25 500 proteins encoded by the Arabidopsis genome, *c.* 34% contain a transmembrane domain and are classified as membrane proteins, and 7% of the 25 500 proteins contain at least four transmembrane domains and could fulfil a transport function (Schwacke *et al.*, 2003; Tang *et al.*, 2020).

Plants, and plant–symbiotic relationships, influence global cycling of carbon, water and nutrients (Silva & Lambers, 2021). The membrane separation processes within plants and plant symbiosomes contribute to the ecosystem services plants provide such as cleaning water, capturing carbon dioxide, cooling land surfaces, moderating the climate and providing food, feed, fibre, fuel and habitats (Fig. 1) (Paasonen *et al.*, 2013; Ellison *et al.*, 2017; Jasechko, 2018; Zhang, 2020). For example, plant transpiration is estimated to be responsible for 60–90% of terrestrial evapotranspiration (Jasechko *et al.*, 2013; Good *et al.*, 2015; Lian *et al.*, 2018; Nelson *et al.*, 2020), plants absorb half of the solar energy that reaches land surfaces, and each year plants capture and fix around one-third of emissions of carbon dioxide.

Demand for the services plants provide is increasing as population growth continues (Maja & Ayano, 2021), but the area of arable land available for plant primary productivity is limited and

at risk of degradation (Praválie *et al.*, 2021). In future, it may be possible to use and remediate land with suboptimal soil composition if plants are selected and engineered to better tolerate challenging soil constraints and climatic conditions (Clemens *et al.*, 2002; LeDuc & Terry, 2005; Rylott & Bruce, 2022). Plant mechanisms for managing excess metals, minerals and salts are also a source of inspiration for developing biotechnological strategies to manage natural resources and waste. This review considers examples of membrane separation mechanisms that plants use to tolerate challenging environments and explores the potential application of these mechanisms in crop engineering and in advancing the development of novel membrane separation technologies. Selected examples of the types of molecules that plants take up and manage, and the types of plant cell membrane mechanisms that are involved in transporting these molecules are included in this review. The examples of molecules and mechanisms included do not constitute a comprehensive list of either the molecules that plants take up or the respective mechanisms involved in the complement of plant separation processes. Future research is needed to identify the additional plant separation mechanisms and to connect the relationships between plant selective mechanism structures and their respective functions in molecular separation.

II. Environment and genetics influence plant ionomes

1. Examples of metals, mineral, nutrient and salt molecules found in soil solutions

Soil solutions across the globe vary in the concentrations of metals, minerals and nutrients that they contain. Some metals and minerals can be toxic to plants when present in excess and there are many nutrients that are essential for plants to draw up because they are required for plant growth (Table 1). The identities of the many molecular mechanisms in plants that enable the uptake and separation of different metals, minerals and nutrients are of interest for improving crop performance and nutrition, phytoremediation strategies, and informing the development of novel membrane biotechnologies (Pilon-Smits, 2005; Schroeder *et al.*, 2013; Mazzolai *et al.*, 2014; Darestani *et al.*, 2019; Gill *et al.*, 2021; Roorkiwal *et al.*, 2021; Kafle *et al.*, 2022).

2. Learning from plant ionomic data

Plant tissue ionomes represent the outcome of plant uptake, separation, partitioning and accumulation of nutrients, metals and minerals (Pita-Barbosa *et al.*, 2019). The ionome refers to the mineral nutrient and trace element composition of an organism (Salt *et al.*, 2008). It is a product of the plant's environment, such as the chemical composition of the soil in which the plant is grown, and the plant's genetics, such as the encoded regulatory systems and mechanisms that influence the uptake of molecules from the soil solution, the plant's developmental stage and regulation of ion transport within the plant (Baxter, 2010; Baxter & Dilkes, 2012). Plant ionomes differ between species, within species and within different plant tissues of any given species. This can be seen in the

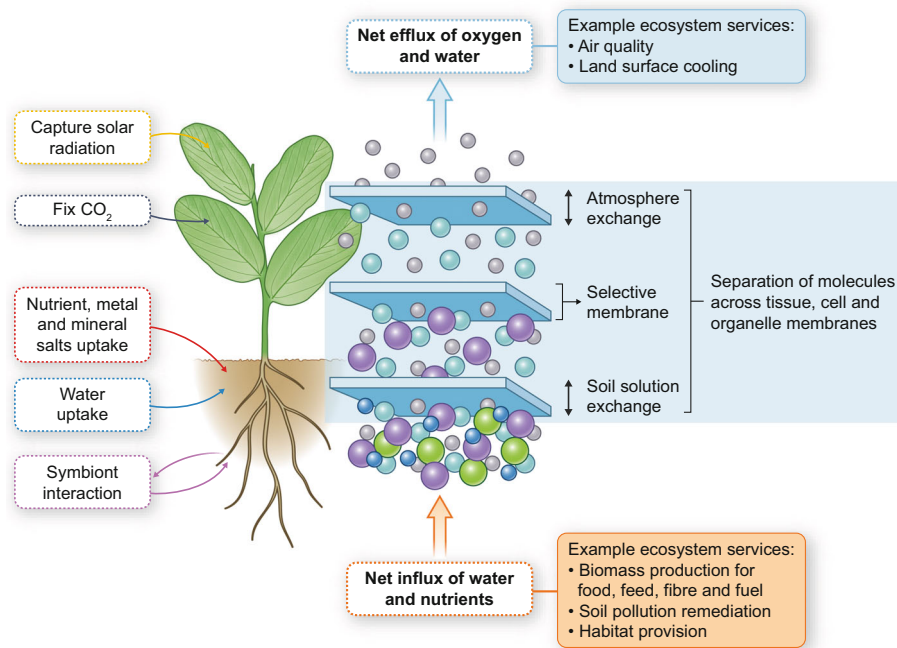


Fig. 1 Plants in collaboration with symbiotic soil microbes contribute services that sustain life. Plants influence global carbon, nutrient (orange dashed box and arrow) and hydrological cycles (blue dashed box and arrows), they fix atmospheric carbon dioxide (CO₂; grey dashed box and arrow), generate net oxygen and clean water (light blue dashed box), capture radiation (yellow dashed box and arrow), cool land surfaces (small blue shaded box) and deliver food, feed, fibre, fuel and habitat (orange shaded box); and this involves exchanges with the soil and atmosphere (large blue shaded box). These services depend on plant photosynthesis, energy conversions, metabolism and membrane separation capability (selective separation of molecules across tissue, cell and organelle membranes). Investigating the molecular mechanisms plants use to achieve their membrane separation functions is key to determining how to improve plant water and nutrient use efficiency, improve crop performance, devise strategies for land remediation and generate biotechnological tools that can be used to address major global challenges such as harvesting resources from wastewater and ensuring food, energy, water and climate security in future.

Table 1 Selected examples of essential nutrient ions plants require (desirable) and ions that share similar charge that can be problematic (undesirable) when present in excess.

	Essential or desirable ions commonly taken up by plants	Ions that can be problematic or undesirable when present in excess
Monovalent cations	K ⁺ , NH ₄ ⁺ , Na ⁺ (for C ₄)	Na ⁺ , NH ₄ ⁺ (in excess), Rb ⁺ , Cs ⁺ , Li ⁺
Monovalent anions	NO ₃ ⁻ , H ₂ PO ₄ ⁻ , H ₂ BO ₃ ⁻ , Cl ⁻	H ₂ BO ₃ ⁻ (in excess)
Divalent cations	Ca ²⁺ , Mg ²⁺ , Co ²⁺ , Cu ²⁺ , Zn ²⁺ , Fe ²⁺	Cd ²⁺ , Pb ²⁺ , Ni ²⁺ , Co(H ₂ O) ₆ ²⁺ , Zn ²⁺
Divalent anions	HPO ₄ ²⁻ , MoO ₄ ²⁻ , SO ₄ ²⁻	CrO ₄ ²⁻
Trivalent cations	Fe ³⁺	Al ³⁺

Abbreviations for ions: aluminium (Al³⁺), ammonium (NH₄⁺), boric acid (H₂BO₃⁻), cadmium (Cd²⁺), calcium (Ca²⁺), cesium (Cs⁺), chloride (Cl⁻), chromate (CrO₄²⁻), cobalt (Co²⁺), copper (Cu²⁺), dihydrogen phosphate (H₂PO₄⁻), hexaaquacobalt(II) (Co(H₂O)₆²⁺), hydrogen (H⁺), hydrogen phosphate (HPO₄²⁻), hydroxide (OH⁻), iron (ferrous cation Fe²⁺, ferric cation Fe³⁺), lead (Pb²⁺), lithium (Li⁺), magnesium (Mg²⁺), molybdate (MoO₄²⁻), nickel (Ni²⁺), nitrate (NO₃⁻), potassium (K⁺), rubidium (Rb⁺), sodium (Na⁺) sulfate (SO₄²⁻), zinc (Zn²⁺). Plants can vary in relation to which elements are essential, for example Na⁺ is required for C₄ plants (Brownell & Crossland, 1972), but not for C₃ plants, although Na⁺ can be useful for osmotic adjustment for all plants (Munns *et al.*, 2020b). Boron, taken up as H₂BO₃⁻, is an example of an essential nutrient that is also toxic when in excess (Raven, 1980), and Zn²⁺ is essential but can inhibit H⁺/OH⁻ channels (Al Khazally & Bielby, 2012). This list provides a collection of examples but it is not a comprehensive list of elements taken up by plants. Plants also absorb other compounds, for example plants absorb urea (CH₄N₂O) and hydrolyse it to ammonium using urease (Witte, 2011), and plants can take up Rare Earth Elements (REEs, see Box 1; Ding *et al.*, 2005). Desirable and undesirable elements and molecules present in soil solutions can share characteristics, such as their size and charge. Distinguishing between different ions that are similar in size and charge is important for ensuring resources and waste are managed efficiently.

data from two studies on *Arabidopsis* ionomes (Fig. 2). In one study, Stein *et al.* (2017) measured the leaf ionome of 1972 *Arabidopsis halleri* samples, paired with the ionome of the soil where each plant grew (Fig. 2a). *Arabidopsis halleri* is a metal hyperaccumulator, able to accumulate levels of zinc, cadmium and

sometimes lead at concentrations an order of magnitude higher than other plants. Stein *et al.* (2017) found that there was large intraspecies variation in the levels of these metals in the plant tissue, which was only partially explained by variation in the soil ion content. This shows that there is intraspecies genetic variation in

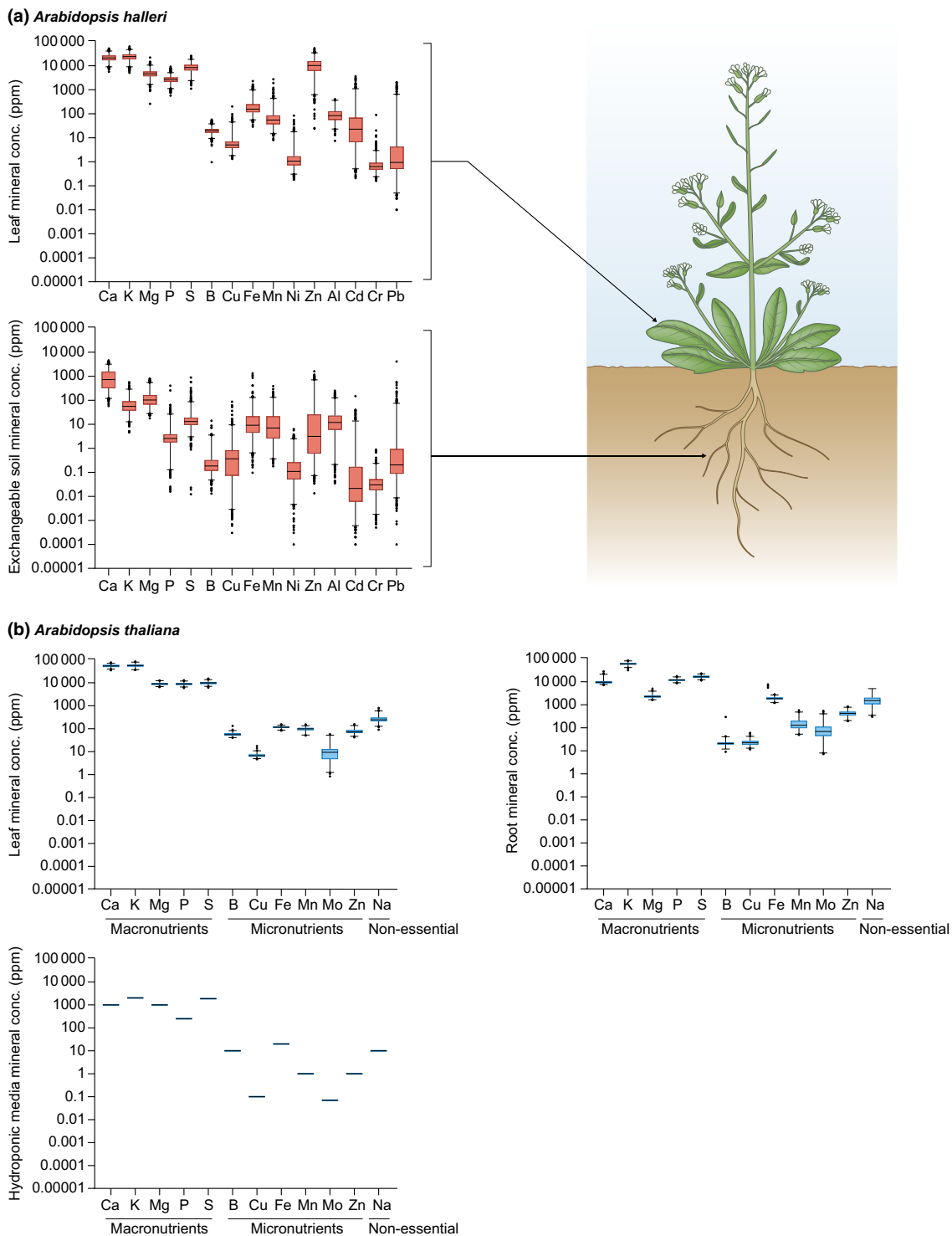


Fig. 2 Element concentrations in *Arabidopsis* tissue samples compared with growth environment. Element concentration values as parts per million (ppm) dry weight are plotted using a box and whiskers graph where the central horizontal line shows the median value, the box shows the 25th/75th percentiles, the whisker bars show 1st/99th percentiles and outliers are plotted as black circles. (a) Concentration of macronutrients, micronutrients and some non-essential elements found in *Arabidopsis halleri* (L.) leaves and the exchangeable mineral concentration of these elements in corresponding soil samples ($n = 1972$). Raw data reported by Stein *et al.* (2017) were plotted using GRAPHPAD PRISM. (b) Concentration of macronutrient, micronutrient and some non-essential elements found in *Arabidopsis thaliana* (L.) leaves, roots and hydroponic growth media ($n = 96$). Raw data reported by Baxter *et al.* (2012) were plotted using GRAPHPAD PRISM (v.8.0.0 for Windows; GRAPHPAD Software, San Diego, CA USA, www.graphpad.com). Abbreviations for units: parts per million (ppm), concentration (conc.), and for elements: calcium (Ca), copper (Cu), boron (B), iron (Fe), magnesium (Mg), manganese (Mn), molybdenum (Mo), potassium (K), phosphorus (P), sodium (Na), sulphur (S), zinc (Zn).

how the ionome is controlled. The authors found that the tissue levels of zinc, cadmium and lead were not strongly correlated despite these metals being chemically similar and sharing some similar transport pathways in plants (Kraemer, 2009). *Arabidopsis halleri*'s ability to differentially accumulate these elements indicates that this species has selective molecular mechanisms capable of differentiating between and separating these similar metals.

The second study, by Baxter *et al.* (2012), measured the leaf and root ionome of 96 *Arabidopsis thaliana* accessions grown in artificial hydroponic media (Fig. 2b). Growing plants in controlled conditions is a useful way to identify how much ionomic variation between plants is due to genetic variation rather than environment. Baxter *et al.* (2012) found that within the measured *A. thaliana* samples, there was variation of greater than an order of magnitude in the levels of some micronutrients – most notably sodium and molybdenum – suggesting intraspecific genetic variation how these elements are managed. There was little correlation between the root and leaf ionomes of the plants. This shows that the ion movement and accumulation within plants is under tissue-specific regulation.

These types of ionomic studies, which allow the identification of relatively extreme ionomic profiles within some types of *Arabidopsis* accessions, have led to genomic analysis and identification of molecular mechanisms associated with differences in ionome profiles (Shariatipour *et al.*, 2021). Similarly, variation in the ionomes of 526 rice (*Oryza sativa*) accessions was used for a genome-wide association study to identify 72 loci associated with ionomic variations. Key candidates included *OsHKT1;5* influencing sodium and *Os-MOLYBDATE TRANSPORTER 1;1* influencing molybdenum (Yang *et al.*, 2018). Clearly, ionomic studies can provide an excellent resource for progressing towards understanding the molecular mechanisms plants use to control element uptake and distribution.

III. Strategies for molecular membrane separation and their biological relevance

Plant membrane separation occurs at the root epidermal cell membrane where the roots meet the soil, across layers of root cortex cells, at passage cells within the Casparian strip, between different plant tissues and between cell cytoplasm and organelle membrane boundaries (Lynch *et al.*, 2021). Cell membranes can separate specific molecules from mixtures because they can have different permeability through the membrane to different molecules. The membrane permeability is influenced by the membrane lipid composition and the embedded proteins in the membrane (Gumbart *et al.*, 2005). Cell membranes contain many different types of proteins such as transporters, pumps and channels. Selective membrane proteins can differentiate between molecules based on the charge number, polarity, size and molecular mass of the molecules (Lepointevin *et al.*, 2017).

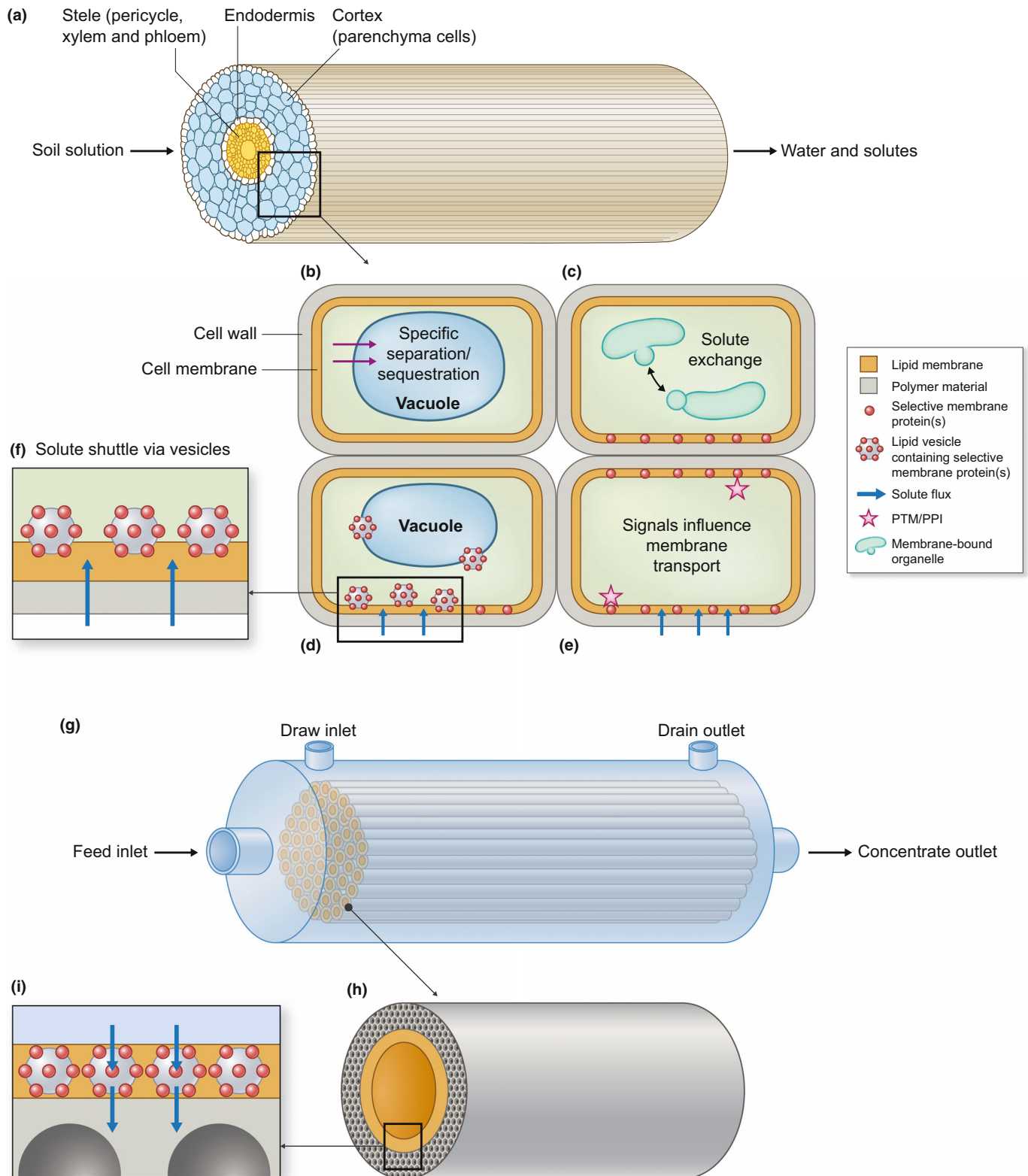
Modular sorting of soluble molecules can occur in plant cells through compartmentation processes, with various compartments acting as reservoirs or repositories of vital molecules or toxic solute dumping sites (Fig. 3). The major cellular compartment is the vacuole, occupying 90% of the volume of plant cells, with membranes packed with specific proteins to enable cellular sorting

and trafficking mechanisms (Tan *et al.*, 2019). Plants modularise separation steps. Molecular mechanisms such as aquaporins that are capable of contributing to bulk solute uptake are used to move relatively large volumes of bulk solutions and this function is coordinated with increasingly selective layering of separation steps.

1. Selective uptake of nutrients

Plants require a minimum of 14 inorganic mineral elements for their growth, development and reproduction, but the soil composition of these essential nutrients can vary widely (Marschner, 2012). Essential nutrients can be classed broadly into macronutrients such as nitrogen (N), phosphorous (P) and potassium (K), which are generally observed in plants at concentrations > 0.1% of plant dry tissue weight, and micronutrients such as iron (Fe), zinc (Zn) and boron (B), generally making up < 0.01% of dry tissue weight (Maathuis, 2009; Samota *et al.*, 2017). Membrane transporters are central to plants taking up nutrients from soil, translocation to shoots and their distribution to different organs and tissues (Sasaki *et al.*, 2016). There are membrane transporter mechanisms with very specific substrate transport roles, and other mechanisms involved in transporting many different substrates, and the nomenclature of different membrane transport mechanisms is often not a useful guide to their role or roles (David *et al.*, 2019; Tyerman *et al.*, 2021). As well as being responsible for the transport and partitioning of nutrients throughout plant tissues, membrane transporters may also have key roles in nutrient sensing (Dreyer *et al.*, 2022). Nutrient sensing is important to ensure the plant can dynamically respond to external nutrient availabilities and trigger additional nutrient transport mechanisms such as vacuolar sequestration or translocation to other tissues. Computational cell biology simulations have suggested that two different transporter types for the same nutrient are needed for homeostatic control (Dreyer, 2021). Therefore, a cell gains flexible control in managing nutrient concentrations through the activity of a combination of transporter types energised by different processes (Dreyer, 2021) (with transporter activity regulation being not linked to differences in their affinity) (Dreyer & Michard, 2020).

Nitrogen is important for plant nutrition The ancestors of land plants, green algae like Characeae, take up ammonium ions (NH_4^+) and methylamine (CH_3NH_3^+) when in need of nitrogen and the rate of transport falls if nitrogen is not needed (Walker *et al.*, 1979). Plants can take up nitrogen (N) from the soil in several forms, such as nitrate (NO_3^-), NH_4^+ and organic molecules (amino acids) (Fan *et al.*, 2017). In some herbaceous plants, the majority of nitrogen is assimilated as NO_3^- which is subsequently reduced to NH_4^+ (Morere-Le Paven *et al.*, 2011). The external concentration of nitrogen can fluctuate greatly in the soil and distinct mechanisms for nitrogen uptake occur in plants depending on the external availability of nitrogen (NO_3^- or NH_4^+) in the soil. Such mechanisms include the use of low-affinity transport systems (LATS), such as the Nitrate Transporter 1 (NRT1) gene family in high nutrient concentrations (mM range), whereas in low nutrient concentrations (μM range), they utilise high-affinity transport



systems (HATS), such as NRT2 transporters, scavenging ions to maintain a normal uptake rate (as reviewed in Fan *et al.*, 2017). Some transporters such as Arabidopsis AtNRT1.1 and *Medicago truncatula* MtNRT1.3, traditionally thought to act in LATS, have

been shown to have dual-affinity properties, with their mode of action dependent on external nitrate concentrations. For example, in high nitrate, AtNRT1.1 functions as a low-affinity nitrate transporter, and in low nitrate it functions as a nitrate sensor, with

Fig. 3 Overview of features of plant and industrial membrane separation structures. (a) Plant tissues, such as the root section shown, are made of files of cell layers where cells are bound by a selective plasma membrane (PM) and supporting cell wall, made of mostly cellulosic or lignocellulosic polymers, along with other cell wall components. Plant vasculature comprised of xylem and phloem vessels connect distant plant tissues to facilitate long-distance solute movement whereas other cell types (e.g. root cortex, pericycle) include many files of cells in each layer. Cell wall structures can influence solute passage but most of the selectivity is determined by mechanisms in the PM. Cells share many common features of membrane function (represented individually in parts b–f), that influence plant cell permeability to different solutes and enable the separation of different molecules, though these examples are not comprehensive or specific (for some specific examples, see: Delsart, 2017; Byrt *et al.*, 2018). (b) Tonoplast-localised membrane proteins influence the transport of molecules in and out of the vacuole. (c) Cytoplasmic streaming influences solute exchange between the cytoplasm and organelles. Endoplasmic reticulum can form plasmodesmal sphincters, potentially enabling flux between cells (Rinne & Schoot, 2004). (d) Plants have intracellular and extracellular vesicles and these vesicles can shuttle solutes, proteins and a range of other molecules (Hwang & Robinson, 2009; Yáñez-Mó *et al.*, 2015; Ueda *et al.*, 2016). (e) Posttranslational modifications (PTMs), protein–protein interactions (PPI) and signalling molecules can influence the localisation of transport mechanisms and alter their functions, including their substrate permeability (Saito & Uozumi, 2020). (f) Plants use vesicles containing membrane proteins, such as aquaporins, for trafficking and intercellular communication (Martinez-Ballesta *et al.*, 2018). Plants also have extracellular vesicles with key roles in moving cargo which are not represented in this diagram (Cai *et al.*, 2021). (g) Representation of the structure of typical hollow fibre membranes. Many systems include outer-selective polymer structures and inner-selective biomimetic thin film composite hollow fibre polyamide membranes, where the membranes consist of thin and dense layers that are porous and selectively permeable (Porter *et al.*, 2020). Different types of hollow fibre membranes are used for a range of applications such as extracting clean water from dirty water, desalination, microfiltration, pervaporation, isotope/gas separation and distillation (Fane *et al.*, 2015; Lau *et al.*, 2022). The specific application determines which liquids move in and out of feed and draw inlets and drain and concentrate outlets (h) The hollow fibres are tube-like consisting of an ultrathin dense selective lipid membrane layer that is mechanically supported on an outer layer of a highly porous substrate of various compositions, where use of organic molecules is ideal because they can be low cost, easily processable and scalable (Sanahuja-Embuena *et al.*, 2019; Lau *et al.*, 2022). (i) The structure in each fibre can include proteoliposomes carrying selective membrane proteins, such as embedded aquaporins which can improve the selectivity, where the proteoliposome lipid vesicles are embedded in a lipid membrane attached to a supporting polymer material (Yılmaz & Özkan, 2022). Polymer materials in separation systems can in some cases contain cellulose similar to the supporting material in plant cell walls. The systems in plants that include cell wall polymer supportive layers, lipid membranes and vesicles containing selective membrane proteins (f) are a source of inspiration for optimising membrane separation technology structures (i). Key: lipid membranes in plants (a–f) and in separation technologies (g–i) are represented in dark yellow; polymer structures and materials are represented in grey; selective membrane proteins are represented as red circles, lipid vesicles containing selective membrane proteins are represented as grey circles containing smaller red circles; the direction of solute flux relative to internal structures is represented with blue arrows (f, d, e, i), whereas black arrows represent the main directions of flow of solutes relative to plant and separation technology structures (a, g); posttranslational modifications (PTMs), protein–protein interactions (PPI) and signalling processes are represented by pink stars; organelles and organelle interactions are represented by green shaded shapes with green outlines; tissues in the stele (pericycle, xylem and phloem) are shaded yellow in part (a); endodermis surrounds the stele; cortex (parenchyma cells) are shaded in light blue in part (a); vacuoles (part b) have dark blue outlines; hollow space for solute flow (in g, h, i) are black.

kinase-mediated AtNRT1.1 phosphorylation resulting in activation of several other high-affinity transporters such as NRT2s (Morere-Le Paven *et al.*, 2011; Rashid *et al.*, 2018). Recently in rice, an aquaporin from the PIP subfamily OsPIP1;3 was reported to facilitate transport of NO_3^- when expressed in HEK cells indicating that aquaporins may also contribute to nitrogen uptake and transport (Liu *et al.*, 2020).

In addition to the regulation of high- and low-affinity transporters, plants are able to modulate the localisation of transporters when nitrogen is in excess. For example, plasma membrane (PM)-localised ammonium transporter AtAMT1;3 is internalised through clustering/endocytosis removing the active transporter from the PM and preventing ammonium uptake in ammonium toxicity conditions (Wang *et al.*, 2013). To adapt to soil constraints, plants may implement polar localisation of AMT1 mechanisms to optimise control of the movement of ammonium, and AMT regulation is adjusted during plant development (Konishi & Ma, 2021; Esmailzadeh-Salestani *et al.*, 2022).

Phosphorus is a finite resource that must be carefully managed Phosphorous (P) is a macronutrient required by all organisms as a building block of nucleic acids and a component of biological membranes. Plants take up P from the soil primarily in the form of inorganic phosphate (Pi). However, P is the least available macronutrient as it easily complexes with metal ions within the soil solution (Shen *et al.*, 2011). To overcome P availability challenges, plants have evolved several strategies for P uptake, and

subsequent translocation throughout the plant (Puga *et al.*, 2017; Prathap *et al.*, 2022). Pi transport is facilitated by multiple P transporters localised on the plasma and intracellular membranes, summarised recently by Wang *et al.* (2021). P concentrations in leaves are highly variable in different plant species (Suriyagoda *et al.*, 2023).

The regulation of Pi homeostasis in plants is complex and involves many mechanisms influencing tissue level, cellular and subcellular regulation of pools of Pi (Kanno *et al.*, 2016). In Arabidopsis, two high-affinity P transporters AtPHT1;1 and AtPHT1;4, localised to the distal root tip, were found to account for up to 75% of Pi uptake (Shin *et al.*, 2004). The transport of P from xylem-to-phloem in Arabidopsis was influenced by a SULTR-like Phosphorus Distribution Transporter (AtSPDT/AtSULTR3;4; Ding *et al.*, 2020). In cereals, the grains are a major P sink (60–85% total plant P) and SPDTs are involved in allocating P to the grain. For example, loss of rice SPDT function resulted in a 20% reduction of P in the grain (Yamaji *et al.*, 2017). A barley (*Hordeum vulgare*) Pi/H⁺ cotransporter HvSPDT, observed to be localised in nodes of reproductive and vegetative tissues, was recently identified as the primary mechanism responsible for distribution of P to grains (Gu *et al.*, 2022). In arsenic hyperaccumulator plants like *Pteris vittate*, there are P transporter encoding genes that were reported to specifically influence P transport, such as *PvPht1;2*, and other P transporter genes encoding mechanisms that transport both P and arsenic, such as *PvPht1;3* and *PvPht1;4* (Cao *et al.*, 2018; Han *et al.*, 2022).

P homeostasis is tightly regulated and influenced by kinase and phosphatases which pause and remobilise P transport, respectively, in response to plant Pi requirements and external Pi conditions (Wang *et al.*, 2021). For example, PHT transporters are subject to multiple levels of post-translational regulation that influence PHT exit from the endoplasmic reticulum (ER), trafficking to the PM, and protein stability in the PM in response to external Pi conditions (Bayle *et al.*, 2011). PHT1 internalised from the PM into endosomes is either recycled in low Pi conditions or targeted for vacuolar degradation in high Pi conditions (Bayle *et al.*, 2011). In rice, the kinase OsCK2 phosphorylates PHT transporters to inhibit their trafficking from ER to PM (Chen *et al.*, 2015) while the protein phosphatase type 2C (PP2C) protein phosphatase OsPP95 interacts with OsPT2 and OsPT8 and dephosphorylates OsPT8 to promote OsPT2 and OsPT8 trafficking from the ER to the PM (Yang *et al.*, 2020).

During Pi starvation, plants increase their capacity to take up Pi and reduce nitrate uptake capacity (Ai *et al.*, 2009). In both Arabidopsis and maize (*Zea mays*), this response is coordinated by a GARP-type transcription factor NITRATE-INDUCIBLE, GARP-TYPE TRANSCRIPTIONAL REPRESSOR1.2 (NIGT1.2). NIGT1.2 upregulates the transcription of Pi transporter genes PHT1;1 and PHT1;4 and downregulates the gene expression of nitrate transporter NRT1.1 under Pi stress conditions (X. Wang *et al.*, 2020). Plants even adjust their mechanisms for managing Pi transport seasonally to account for seasonal changes in Pi availability (Kurita *et al.*, 2022).

Potassium is essential for all living cells Potassium (K⁺) fulfils many critical functions in plants from osmotic balance maintenance, regulation of cell membrane potential to driving transport processes and metabolic processes, and K⁺ serves as an enzymatic cofactor and influences turgor generation for cell expansion and guard cell aperture control (Dreyer & Uozumi, 2011; Srivastava *et al.*, 2020). In typical soil conditions, the K⁺ concentration is between 10 and 100 μM, whereas the optimal K⁺ concentration in plant cell cytoplasm ranges between 100 and 200 mM (Wyn Jones & Pollard, 1983; Schroeder *et al.*, 1994; Sharma *et al.*, 2013); to maintain required K⁺ concentrations in cells plants employ a complex K⁺ transport system (Ragel *et al.*, 2019). There are two major pathways for K⁺ uptake and translocation, these are also often pathways for Na⁺; one is mediated largely by selective ion transporters, and the other is mediated largely by ion channels (Britto & Kronzucker, 2008). Different pathways may be active under different external K⁺ conditions and different mechanisms can have different K⁺ affinities (Nieves-Cordones *et al.*, 2014).

Similar to nitrogen, potassium is taken up from the soil by the two-mechanism model of HATS and LATS and the different mechanisms have distinct transport energetics (Britto & Kronzucker, 2008). Under high soil K⁺ concentrations (≥0.5 mM), K⁺ uptake is predominately carried out by low affinity inwardly rectifying K⁺ channels (e.g. AKT1; Hirsch *et al.*, 1998), whereas under low K⁺ concentrations (below 0.1 mM) high-affinity transporters are active (e.g. HAK1 and HAK5; Grabov, 2007). Potassium channels in plants are generally either voltage-gated (e.g. AKT1) or voltage independent (e.g. TPKs; Isayenkov *et al.*, 2011)

and can be specific for K⁺ or non-selective (Britto & Kronzucker, 2008; Dreyer & Uozumi, 2011). Non-selective cation channels (NSCCs) is the umbrella term describing transmembrane channel proteins found in both animal and plant cells that are generally permeable to a wide range of cations but with a heterogeneity of other kinetic, energetic and regulatory characteristics (Demidchik & Maathuis, 2007). NSCCs have been implicated in both K⁺ and Na⁺ flux in roots. Under normal conditions, NSCCs function primarily in K⁺ uptake and transport systems (Pottosin & Dobrovinskaya, 2014); however, root electrophysiological studies suggest that NSCCs also facilitate the majority of Na⁺ influx in saline conditions (for review, see Kronzucker & Britto, 2011). There are three major K⁺ transporter families in plants, the KT/HAK/KUP, the TRK/HKT and the CPA cation proton antiporter families and members of these transporter families generally make up the HATS that operate under normal or low K⁺ conditions (reviewed in Gierrth & Mäser, 2007). In Arabidopsis roots, HAK5 and AKT1 are the primary mechanisms for K⁺ uptake from the soil, mediating the majority of K⁺ absorption (Gierrth *et al.*, 2005; Pyo *et al.*, 2010). However, the full complement of potassium transport mechanisms is not yet known. When both AKT1 and HAK5 were knocked out in Arabidopsis as the two primary K⁺ uptake transport proteins, an unknown mechanism contributed to K⁺ uptake under high K⁺ conditions (Rubio *et al.*, 2010); recent reports of K⁺ permeable ion channel aquaporins (Qiu *et al.*, 2020; Tran *et al.*, 2020) reveal that these channels could be contributing to K⁺ nutrition (Tyerman *et al.*, 2021), although this is yet to be confirmed *in planta*.

Potassium transporters are tightly regulated to ensure exquisite control over cell potassium homeostasis from nutrient sensing leading to transcriptional changes, to post-translational modification or protein heterotetramerisation to enable modulation of transport activity (reviewed in Wang & Wu, 2013, 2017). The gene expression of many K⁺ transporters is regulated in response to K⁺ deficiency (Ashley *et al.*, 2006), for example high-affinity K⁺ transporter HAK5 is transcriptionally upregulated in response to low K⁺ conditions under the control of a MYB transcription factor MYB77 (Feng *et al.*, 2021). The activity of several K⁺ transporters has been reported to be regulated by phosphorylation (Wang & Wu, 2013), often in response to low K⁺ stress. For example, under K⁺ deficit, two Ca²⁺ sensors, AtCBL1 and AtCBL9, activate the kinase CIPK23 which phosphorylates AtAKT1 to increase its activity (Xu *et al.*, 2006). Activation by phosphorylation has also been reported for HAK5 (Ragel *et al.*, 2015; Scherzer *et al.*, 2015), and stomatal outwardly rectifying K⁺ channel GORK (Van Kleeff *et al.*, 2018). Transport activity inhibition by dephosphorylation mediated by phosphatases was reported for GORK (Lefoulon *et al.*, 2016) and for AKT1 (Lee *et al.*, 2007). The activity of AtAKT1 is also negatively regulated by its heterotetramerisation with another K⁺ channel subunit AtKC1 (Duby *et al.*, 2008; Wang *et al.*, 2010).

Potassium homeostasis is intricately linked to plant nutrient status, salt stress tolerance and cell energy status. There is emerging evidence of integrated control between K⁺ and N nutrition (reviewed in Ruffel, 2018, Srivastava *et al.*, 2020). For example, expression of N transporters was significantly affected by deficient

K^+ conditions and in *Arabidopsis* a low-affinity NO_3^- transporter (NRT1.5) was found to function as a H^+/K^+ antiporter, effluxing K^+ from the root parenchyma cells to the xylem (Li *et al.*, 2017). The maintenance of a high cytosolic $K^+ : Na^+$ is widely considered a major determinant of salt tolerance (Munns & Tester, 2008) and is contributed to by K^+ and Na^+ channels and transporters alike (see *Salt stress*, Amtmann & Beilby, 2010). For example, a positive correlation was found between K^+ uptake and salt tolerance in several wheat (*Triticum aestivum* L.) cultivars (Cheng *et al.*, 2015). Halophytic ('salt tolerant') plants can be characterised by their ability to maintain sufficient $K^+ : Na^+$ within their cells (Shabala & Cuin, 2008; Flowers *et al.*, 2015) tolerating much higher Na^+ tissue concentrations through the use of several membrane separation mechanisms and in general a higher selectivity for K^+ uptake (Flowers & Colmer, 2008). Potassium is a molecule of importance related to cell energy regulation. For example, a 'rechargeable K^+ battery' has been described, which can influence sucrose loading in phloem under conditions where H^+ -ATPase activity is limited; and cycling of K^+ in phloem and surrounding cells via the K^+ channel ATK2 can generate electro-chemical gradients to energise the PM for other transport processes (Gajdanowicz *et al.*, 2011; Dreyer *et al.*, 2017). If plants are struggling to maintain phloem pressure due to sucrose being limited K^+ can be used to help adjust hydrostatic pressure, supporting flow in the phloem (Babst *et al.*, 2022).

Resource separation involves coordinated regulation of mechanisms varying in selectivity Plants have many complex mechanisms for nutrient uptake and translocation, but the membrane separation systems for managing target molecules can be imperfect with respect to their selectivity. Where there are undesirable toxic solutes chemically similar to desirable nutrients, the undesirable solutes can be taken up and translocated throughout the plant inadvertently. Examples of this include excess Na^+ uptake via systems for K^+ uptake (Kronzucker & Britto, 2011, and above), and communal mechanisms for silicon and arsenic uptake (Ma *et al.*, 2008; Zhao *et al.*, 2009). Silicon, a beneficial solute, for most plant species particularly in stress response (for review and references within see: Coskun *et al.*, 2019), is taken up from the soil solution and translocated throughout the plant by a cooperative and imperfectly selective membrane transport system (Ma & Yamaji, 2006, 2015; Ma *et al.*, 2007). This cooperative transport system was first identified and characterised in rice, where rice requires silicon to maintain stable and high yields, but homologues of the transporters have since been reported in several other species (e.g. barley Yamaji *et al.*, 2012; Chiba *et al.*, 2009), maize (Mitani *et al.*, 2009a,b), tomato (*Solanum lycopersicum* L; Mitani-Ueno & Ma, 2021) and cucumber (*Cucumis sativus*; Sun *et al.*, 2017).

In rice, silicon transport is facilitated by two aquaporin channel proteins Lsi1 (OsNIP2;1; Ma *et al.*, 2006) and Lsi6 (OsNIP2;2; Yamaji *et al.*, 2008) localised to the root and shoot, respectively, and two known active transporters Lsi2 (ion transporter superfamily; Ma *et al.*, 2007) and Lsi3 (ion transporter superfamily; Huang *et al.*, 2022) localised to root tissue though different cell types. In roots, Lsi1-, Lsi2- and Lsi3-mediated silicon uptake occurs through polar localisation and specialised transport activity; neutral silicic

acid is passively transported into root cells from the external environment via the channel protein aquaporin Lsi1 (NIP2;1); then active efflux transporters Lsi2 and Lsi3 mediate silicon loading into the xylem for root-to-shoot translocation. However, both Lsi1 and Lsi2 are also known to facilitate the uptake, transport and translocation of the toxic solute arsenic (Ma *et al.*, 2008; Konishi *et al.*, 2022). Mutations to both Lsi1 and Lsi2 significantly decreased arsenic uptake and its accumulation in shoot and grain tissues (Ma *et al.*, 2008). Improvements to silicon transporter selectivity through protein engineering could be an approach to retain the benefits of silicon nutrition to plant growth and yield while minimising toxicity impacts of arsenic to both plants and humans (Clemens, 2019).

2. Dealing with stress

Environmental stress has been a selective pressure driving plants to evolve specialised molecular separation mechanisms. Element toxicity is one type of stress that limits plant growth and yield (Dudka & Miller, 1999). Depending on the species, element toxicity can also have major human health impacts, for example arsenic accumulation in rice (Zhao *et al.*, 2010; Clemens & Ma, 2016). In plants, element toxicity disrupts essential cellular processes such as photosynthesis, protein synthesis and enzymatic reactions as a consequence of the over-accumulation of some elements (Singh *et al.*, 2016). Plants employ several detoxification strategies to manage high toxic element concentrations that rely on the separation of the toxic species across membranes, a process which is facilitated by transport proteins (Volkov & Beilby, 2017). Strategies to manage the distribution of toxic elements within cells and between the shoots and roots include the following: (i) restriction of influx into the cell or translocation throughout the plant (exclusion and retrieval); (ii) the export of excess ions, such as from the root to the rhizosphere (extrusion); and (iii) compartmentation and/or hyperaccumulation of these ions into vacuoles.

Salt stress Salinity is a geographically widespread example of a soil constraint that can cause toxicity problems for plants, and salinisation of agricultural soils is increasing globally (Hassani *et al.*, 2021). Excess salt concentrations in soil solutions can result in cellular toxicity problems, where Na^+ tends to be the toxicity causing element in many plant species, and Na^+ influences whether plants can maintain a suitable gradient for water and nutrient uptake and retention (Jones & Gorham, 2002). The primary strategy for crop plant salt stress tolerance is the maintenance of low intracellular Na^+ concentrations, whereas halophytic plants can accumulate and sequester salt ions (Flowers *et al.*, 2015; Van Zelm *et al.*, 2020).

Crop plant roots must exclude 97–98% of the Na^+ and Cl^- present in saline soils or these ions would quickly accumulate to toxic levels (Munns *et al.*, 2020b). Membrane separation mechanisms circumvent/prevent toxic accumulation of Na^+ by direct Na^+ extrusion from the root epidermis into the soil environment, retrieval of Na^+ from the xylem sap in the root to prevent root–shoot translocation and shoot Na^+ accumulation, and sequestration of Na^+ into cell vacuoles (for recent review, see: Van

Zelm *et al.*, 2020). Several transporters have been identified as contributing to these mechanisms of salt stress tolerance and adaptation.

The Salt Overly Sensitive (SOS) pathway constitutes a key mechanism for salt tolerance through sodium extrusion from the plant into the rhizosphere contributing to the maintenance of low cellular Na^+ concentrations (Ji *et al.*, 2013). The pathway involves salt stress sensors, signalling intermediates and regulatory proteins (Liu & Zhu, 1998; Halfter *et al.*, 2000) that coordinate the control and downstream activity of the PM-localised NHX-family Na^+/H^+ antiporter SOS1 (Lin *et al.*, 2009; Quintero *et al.*, 2011). The SOS1 transporter is activated by phosphorylation mediated by the SOS2/SOS3 kinase complex (Qiu *et al.*, 2002) and works in combination with PM- H^+ ATPase to maintain a sufficient proton gradient (Gévaudant *et al.*, 2007; Bose *et al.*, 2015; Miranda *et al.*, 2017) so that the pathway functions as a Na^+ exclusion module which has been observed in many glycophyte and halophyte species. SOS1-mediated Na^+ extrusion from the root-to-rhizosphere was reported for epidermal cells of the root tip (Shi *et al.*, 2002). Meristematic cells lack large vacuoles and are not connected to vascular tissues which limits the type of mechanisms they can deploy to manage excess Na^+ . The overexpression or knock-out (KO) of SOS1 has been shown to influence salt tolerance in Arabidopsis (Shi *et al.*, 2003; Yang *et al.*, 2009), tobacco (*Nicotiana tabacum*; Yue *et al.*, 2012), rice (El Mahi *et al.*, 2019) and maize (Zhou *et al.*, 2022). However, the activity or lack thereof of SOS1 does not always confer salt tolerance or sensitivity respectively; when *sos1* loss of function Arabidopsis experienced mild salt stress (25 mM NaCl), these plants accumulated less shoot Na^+ compared with wild type (WT), whereas under severe salt stress (100 mM NaCl), *sos1* mutants accumulated more shoot Na^+ compared with WT (Shi *et al.*, 2002). Similarly, *sos1* mutants were reported to have reduced accumulation of Na^+ , within the external $[\text{Na}^+]$ range of 5–75 mM (Ding & Zhu, 1997).

There is limited evidence that the salt tolerance of halophytic species is related to more efficient or different SOS1 activity compared with glycophytes, and salt tolerance in halophytic species is not always associated with an increased rate of Na^+ efflux (Britto & Kronzucker, 2015). SOS1 has been identified in the halophytic relative of Arabidopsis *Eutrema salsuginea* (previously known as *Thellungiella salsuginea*) and knock-down of SOS1 resulted in a salt-sensitive phenotype, and Na^+ accumulated preferentially in the root xylem parenchyma cells displacing cytosolic K^+ (Oh *et al.*, 2009). Halophytism of *E. salsuginea* was attributed to increased SOS1 expression rather than hyper-functioning SOS1 (Oh *et al.*, 2009), but a comparative study between Arabidopsis and *E. salsuginea* found that Na^+ influx in the root was significantly less in *E. salsuginea*, whereas Na^+ efflux was greater in Arabidopsis (Wang *et al.*, 2006). In halophytic *Salicornia spp.*, SOS1 was observed to have constitutively high expression (Yadav *et al.*, 2012; Katschnig *et al.*, 2015). High salinity can sometimes impair the proton pumping that is needed to power Na^+/H^+ antiport, and some organisms exhibit Na^+ efflux powered directly by Na^+ -ATPase activity, for example in salt-tolerant Characeae, some bryophytes and *Tetraselmis algae* (Whittington & Bisson, 1994; Balnokin *et al.*, 1997; Benito & Rodríguez-Navarro, 2003; Matalin

et al., 2021; Phipps *et al.*, 2021a,b; Beilby *et al.*, 2022). Rice lines capable of producing greater biomass in saline conditions were created previously by expressing a *Physcomitrella patens* Na^+ -ATPase indicating that there is scope to further explore how manipulating Na^+ -ATPase function influences plant performance in saline conditions (Jacobs *et al.*, 2011). The contribution of SOS1 to salt tolerance in general and to overall plant energetics has been given consideration in previous studies. If extrusion of one Na^+ from the root by SOS1 requires one H^+ to be similarly extruded by H^+ -ATPase, based on the root Na^+ effluxes measured (or calculated) this would consume more energy in the form of ATP than is produced by the maximal respiration of the root (Britto & Kronzucker, 2009; Munns *et al.*, 2020a). This inefficiency highlights that the use of SOS1 is only one option among a range of mechanistic options contributing to salt stress response and management of Na^+ transport in plants.

The retrieval of Na^+ from root xylem to surrounding cell layers, enabling control of Na^+ translocation to the shoot to reduce shoot Na^+ accumulation, is a well-characterised mechanism contributing to salt stress adaptation and tolerance (Assaha *et al.*, 2017). The High-Affinity Potassium Transporter (HKT) protein family has been reported to contribute to this type of Na^+ retrieval mechanism (Hamamoto *et al.*, 2015). There are HKTs with roles in K^+ nutrition and HKT isoforms capable of Na^+ uniport (Class I), some HKTs contribute to Na^+ and K^+ symport (Class II – present only in monocots; and rice HKT2;1 is class II but is a Na^+ transporter: Horie *et al.*, 2001) (Riedelsberger *et al.*, 2021). HKT function has been linked to salt tolerance traits in many species (including halophytes: Ali *et al.*, 2021), for example HKT1;5 function in wheat (Byrt *et al.*, 2007) and rice (Ren *et al.*, 2005) was associated with salt tolerance. HKTs possess a selectivity filter located at the pore entry G/S-G-G-G where the difference in selectivity between Class I and Class II HKTs has been attributed to this serine–glycine sequence polymorphism (Mäser *et al.*, 2002), and this structural feature is therefore an important determinant of function in salt tolerance mechanisms. Recent studies have also identified additional residues influencing HKT function and conferral of salt tolerance. In HvHKT1;5, a leucine to proline substitution at position 189 (L189P) was found to perturb its PM localisation and Na^+ transport function (Houston *et al.*, 2020). Similarly, two HKTs identified in major shoot Na^+ exclusion loci in wheat *Triticum monoccocum* and *Triticum aestivum* differed in their Na^+ affinity through a single amino acid difference (Xu *et al.*, 2018). Engineering of higher Na^+ -affinity HKTs, combined with their cell type specific overexpression (Møller *et al.*, 2009), can influence Na^+ retrieval, Na^+ shoot exclusion and plant salt tolerance.

High-Affinity Potassium Transporters work in concert with other mechanisms. A HKT isoform previously identified in maize as functioning in Na^+ retrieval from the xylem was recently found to work cooperatively with a Na^+ selective transporter from the HAK family. ZmHAK4 was identified by GWAS as conferring variation in shoot Na^+ exclusion and salt tolerance in maize plants, accounting for c. 11% of shoot $[\text{Na}^+]$ variation (Zhang *et al.*, 2019). ZmHAK4 is preferentially expressed in the root stele and xylem parenchyma cells, localised to the PM, and is Na^+ selective (Zhang *et al.*, 2019). Single ZmHAK4 and ZmHKT1 CRISPR KO lines

differed in their root and shoot Na^+ content. The ZmHAK4 KO had greater shoot but lower root Na^+ content than WT under all test conditions, whereas the ZmHKT1 KO differed from WT under extreme conditions (120 and 150 mM NaCl) only. This suggests that this cooperative mechanism acts under different salinity conditions, where ZmHAK4 is a primary mechanism for Na^+ root to shoot translocation under relatively low Na^+ conditions (up to 50 mM) (despite having lower Na^+ affinity than HKT1, HAK4 transcript levels were 35-fold higher), whereas HKT1 is a key mechanism used under relatively high Na^+ conditions. Dual-affinity mechanisms are also described for nutrient uptake (See *Selective uptake of nutrients*).

Another mechanism that plays a central role in plant salt tolerance is Na^+ compartmentalisation within cells. Once inside the plant cell, compartmentation of Na^+ ions can occur through vacuolar sequestration by transporters localised in the tonoplast (vacuole membrane). A key transporter class example is the vacuolar Na^+/H^+ antiporters (NHXs, Group I); their Na^+ sequestration activity has been shown to increase in response to salt stress, reviewed by Horie & Schroeder (2004) and Keisham *et al.* (2018).

Selective Na^+ compartmentation can also extend to a 'tissue-level' accumulation, for example several grass species such as wheat, barley and sorghum (*Sorghum bicolor*) have been shown to load excess Na^+ into older tissues, such as old blades and leaf sheaths, to protect young tissues from toxicity upon salt stress (Wei *et al.*, 2003; Netondo *et al.*, 2004; Huang *et al.*, 2006; Rahnama *et al.*, 2011).

Halophytic ('salt tolerant') plants can withstand higher and longer exposure to salt stress than glycophytic plants. Halophytes are capable of osmotically adjusting to maintain a positive turgor pressure in soils containing more than 200 mM NaCl. These plant types have developed improved transporter-mediated cellular uptake and compartmentation of Na^+ and Cl^- ions to ensure monovalent ions concentrations are maintained within tolerable limits, reducing toxicity (Cheeseman, 1988; Jones & Gorham, 2002; Hao *et al.*, 2021). A subset of halophytes (also known as cretahalophytes) are capable of not only compartmentalising Na^+ ions intracellularly, but also secreting excess salt from their leaves through specialised cellular (i.e. salt bladder) or multicellular structures (i.e. salt gland) (Flowers *et al.*, 2015). Salt-secreting structures can vary in cell number and complexity, with dicotyledonous plants generally having more complex multicellular structures sunken in the leaf epidermis, and mangrove species such as *Avicennia marina* having combinations of 8–12 collecting and secretory cells within the salt gland structure (as reviewed by Yuan *et al.*, 2016). Proteomics studies identified a multitude of PM (e.g. NHX and ATPases) and tonoplast-localised (Potassium transporters and V-type proton ATPase) transporters associated with Na^+ secretion through mangrove salt glands (Tan *et al.*, 2015). Identified transporter candidates also included aquaporin channel proteins belonging to both the PM- and tonoplast-localising subfamilies (PIPs and TIPs, respectively), potentially implicating them in regulating water and ion transport at both these cellular locations (Natarajan *et al.*, 2021; Guo *et al.*, 2022). Aquaporins may contribute to mechanisms in halophytes, such as mangroves, that enable them to thrive in

Box 1 Plants can take up and accumulate rare earth elements (REEs).

This feature is of interest for phytoextraction of these types of resources and for engineering membrane separation technologies (Bashiri *et al.*, 2022; H. Chen *et al.*, 2022; Okoroafor *et al.*, 2022). Rare earth elements (REEs) include holmium (Ho), erbium (Er), thulium (Tm), ytterbium (Yb), lutetium (Lu), gadolinium (Gd), yttrium (Y), terbium (Tb), dysprosium (Dy), neodymium (Nd), europium (Eu), lanthanum (La), cerium (Ce), praseodymium (Pr), promethium (Pm), samarium (Sm) and scandium (Sc) (De Boer & Lammertsma, 2013). Plants have been observed to fractionate different REEs to different tissues (Ding *et al.*, 2005).

osmotically challenging environments, such as growing in seawater (Coopman *et al.*, 2021). Resolving how plants achieve control over separating salt and water molecules can contribute to technological advances, such as inspiring novel desalination technologies (Kim *et al.*, 2016; Y. Wang *et al.*, 2020).

Metal toxicity Among the many different types of metal and element ions, some are useful for plants and others are not. For example, metal ions such as Co, Cu, Fe, Mn, Ni and Zn are required as co-factors for executing diverse enzymatic reactions essential for plant growth, whereas others such as As, Cd and Pb have no known functions for plant growth. Although some metals are essential for cell functioning, excess concentrations can result in toxicity due to cellular stress and oxidative damage. Therefore, sorting of desirable vs undesirable metals and elements is essential for plants' survival (see Table 1 and Box 1).

To deal with the uptake of heavy metals from the soil, plants can distribute heavy metals to different target areas, such as deposition in cell walls or sequestration in vacuoles. Within cells, cytoplasmic ligand binding is also an important survival strategy, enabling long-distance transport of metals and decreasing metallic toxic effects in the cytoplasm (recently reviewed in Jogawat *et al.*, 2021).

There are a multitude of strategies occurring in plants which present a coordinated approach to sort, redistribute and sequester metals through membrane separation processes, specifically using membrane transporters for cellular and organellar redistribution of metals (Kim *et al.*, 2002, 2006, 2009; Song *et al.*, 2003, 2010a,b; Lee *et al.*, 2005; Li *et al.*, 2022). Studying these mechanisms has the potential to inform the development of novel technologies for separating metals (Fig. 3).

Plants can perform modular separation across various cellular membranes. The PM is the first membrane barrier molecules cross when moving into the cell, and the PM of root epidermal cells may benefit from using less selective transporters because this helps in facilitating the entry of a range of metals into the cell where they can then be sorted. Plasma membrane-localised and poorly selective cation transporters such as ZRT/IRT-like (ZIP) have broad substrate specificity for a multitude of heavy metals such as Zn, Co, Cd, Mn and Ni (Korshunova *et al.*, 1999). ZIPs can therefore provide a low-specificity uptake mechanism for heavy metals, where subsequent more precise separation is achieved by the use of

other transporter types, such as specific vacuolar sequestration mechanisms. For example, transporters in the Iron-regulated Protein (IREG) family mediate specific vacuolar sequestration of Ni inadvertently taken up by IRT1 in roots (Schaaf *et al.*, 2006).

Further strategies adopted by plants to deal with metal ions include the collaboration of membrane-associated signalling molecules, calcium-dependent protein kinases (CPK) and metal transporters, with CPKs acting as sensors catalysing the activation/inactivation of specific transporters to control selective separation across membranes. For example, CPK31 interacts with AtNIP1;1, an aquaporin involved in As uptake, where CPK31 activates AtNIP1;1-mediated As uptake (Ji *et al.*, 2017).

Within the plant kingdom, there are plants which can grow in highly metalliferous soils, accumulating relatively high metal concentrations *in planta*. Hyperaccumulator plants can sequester toxic heavy metal elements *in planta*, without the plant showing toxicity symptoms. Such species are able to bioaccumulate heavy metals such as Ni, Zn, Cd, Pb, Co, Cu, Mn, Cr, Se and As, to concentrations 100–1000 times higher than other plants (van der Ent *et al.*, 2013; Lange *et al.*, 2017). Hyperaccumulation strategies have evolved multiple times in the plant Kingdom, as such there might be different strategies adopted by plants to survive metalliferous soils conditions with some species capable of hyperaccumulating more than one element (Reeves *et al.*, 2018). Different compartmentation strategies have been observed in plants for dealing with specific metals. For example, in dealing with Cd, hyperaccumulator species *Thlaspi caerulescens* partitions 35% Cd in the cell wall and remainder in cells/vacuoles (Coso *et al.*, 2005), whereas tobacco plants accumulate virtually all Cd in vacuoles (Vogeli-Lange & Wagner, 1990). In the Cd/Zn hyperaccumulator, *Sedum plumbizincicola* HMA3 was involved in conferring Cd tolerance (Liu *et al.*, 2017). HMA3 mechanisms are also relevant in cereals. For example, a Sorghum ionomics study and QTL analysis identified Cd transporter SbHMA3s as a key mechanism controlling low Cd seed accumulation by sequestering excess Cd in roots, and *OsHMA3* influences Cd accumulation in rice, and overexpression of rice *OsHMA3* in wheat decreased Cd accumulation in wheat grains (Ueno *et al.*, 2010; L. Zhang *et al.*, 2020; Wahinya *et al.*, 2022).

Hyperaccumulator plants can be used to extract metals from metal-rich soils and wastes, a process termed phytomining or agromining (Brooks *et al.*, 1998; van der Ent *et al.*, 2013, 2015). In the order of 0.2%, angiosperms can hyperaccumulate heavy metal ions (Baker & Whiting, 2002). Metals may be harvested from the plant biomass through smelting to produce bio-ores (Novo *et al.*, 2017). Phytomining to capture metals from wastes is considered to be a strategy of growing importance to transitioning to a circular economy (Dinh *et al.*, 2022). A growing number of studies are emerging that assess the potential of different plant species to contribute to phytomining. For example, a list of plant species that are of interest for harvesting Au, Ag, Pd, Pt and Rh were tabled by Dinh *et al.* (2022) in an analysis of phytomining processes and opportunities, and the gold hyperaccumulation in 17 different plant species was compared by Kurniawan *et al.* (2022) revealing species such as *Typha angustifolia* and *Cyperus haspan* growing around gold mining tailings dams that can accumulate

0.56 mg kg⁻¹ gold in dry biomass. The range in gold accumulation in plant shoot samples from the 17 species was 0.44–0.63 mg kg⁻¹ and the authors estimate that the most notable species with phytomining potential, *Typha angustifolia* and *Cyperus haspan* could potentially yield in the order of 2.9 g ha⁻¹ Au dry weight (Kurniawan *et al.*, 2022).

Plant interactions with soil microorganisms and herbivores influence the capability of plants to take up and manage metal extraction and the potential benefits of hyperaccumulation, respectively (Zubair *et al.*, 2016). For example, rhizobacteria can influence the availability of metals like nickel in the soil. Previous studies reported that the presence of *Microbacterium arabinogalactanolyticum* increased *Alyssum murale* nickel uptake into the shoot by 32.4% (Abou-Shanab *et al.*, 2003). *A. murale* can accumulate Ni and Co to above 1000 µg g⁻¹ dry weight (Tappero *et al.*, 2007). Hyperaccumulation of some metals like nickel or arsenic may be an advantage to some plants in relation to preventing herbivory (Palomino *et al.*, 2007). For example, arsenic accumulation in the fern *Pteris vittate* was reported to limit grasshopper feeding damage (Rathinasabapathi *et al.*, 2007). *Pteris vittate* can accumulate several 1000 mg as per kg (Zhenyan, 2022). An SLC13-like AsIII effluxer called PvAsE1 is implicated in As translocation via xylem loading contributing to *P. vittate* As hyperaccumulation (Zhenyan, 2022). The mechanisms plants have evolved to manage soils with challenging metal content are relevant to informing the design of systems for separating metals from metal-rich wastewater sources.

3. Membrane structures and regulation

Controlling membrane metal, mineral and nutrient permeability is important for all cellular life and it is also important for developing precision membrane separation technologies (Mulikidjanian *et al.*, 2009). Living cells can adjust cell and organelle metal, mineral and nutrient content and transport through processes such as vesicle transport, changing inter-organelle dynamics and the form of substrate molecules, and by controlling membrane protein functions via post-translational modification (Fig. 3).

Plants have heterogeneous populations of intracellular and extracellular vesicles, which are lipid bilayer-enclosed spheres that have roles in exchanging molecules between organelles, cells, tissues and the external environment including exchange of molecules with other organisms such as microbes (He *et al.*, 2021; Urzi *et al.*, 2021). Combining vesicular membrane structures into membrane separation technologies has aided technological development (Perry *et al.*, 2015). For example, vesicles with embedded aquaporins are attached to polymer layers on porous support substrates to optimise forward and reverse osmosis membrane technology functions (Perry *et al.*, 2015; Fig. 3).

One of the strategies living cells use to influence membrane permeability to important substrates is to convert the form of the substrates following transport to help maintain favourable electrochemical gradients. This can be achieved by changing the form of the substrate during or directly after the membrane transport step. Some types of membrane transport mechanisms are all-in-one units for transport and substrate conversion but other mechanisms

require multiple parts, such as transporter-enzyme complexes, to achieve the transport and conversion of substrates. For example, some types of plant channels can add or remove protons from ammonia or ammonium ions, respectively, during the transport of these substrates across cell membranes (Kirscht *et al.*, 2016). Whereas for molecules such as carbon dioxide living cells pair transport components with enzymes such as carbonic anhydrase to enable them to switch the form of molecule between carbon dioxide and bicarbonate (Reithmeier, 2001). In the field of advancing membrane separation capability, there has been limited investigation of the potential for using diverse plant transporter-enzyme complexes to inform the development of novel components for precision separation applications.

Native cell environment influences the structure and function of membrane proteins (Robertson, 2018). This means that the endogenous function of a protein may not always be replicated when the protein is present in a different membrane environment, but many membrane transport mechanisms can retain function when expressed in an alternative system, such as a heterologous expression systems or artificial membranes (Opekarova & Tanner, 2003). For example, expression of plant metal transporters in yeast cells was reported to increase yeast cell metal accumulation. Engineered yeast strains expressing plant metal transporter and tracking pathway mechanisms caused yeast to sequester metals at concentrations 10–100 times more than established thresholds for chromium, arsenic and cadmium (Sun *et al.*, 2019). Membrane proteins are also subjected to testing following reconstitution in lipid bilayers assuming that their function in their endogenous environment can be recreated in an alternative membrane environment (Kimelberg, 1976). Many membrane proteins have features that living cells can use to turn them on and off, and to alter their permeability to different substrates (Schönichen *et al.*, 2013; Qiu *et al.*, 2020). For the purpose of advancing separation technologies, adjustment of membrane protein functional switches, such as residues subject to post-translational modifications, may contribute to providing consistency in the permeability properties of the proteins.

IV. Bioengineering selective membrane separation technologies

Molecules that are important for plants are important for food and energy security. Half of the global population depends on fertiliser nitrogen for access to sufficient food (Dawson & Hilton, 2011). Urea and ammonium are important sources of N for plants and important molecules for the emerging hydrogen economy (Lan *et al.*, 2012), and urea is an important additive influencing diesel engine emissions (Demir *et al.*, 2022). P and Fe are essential plant nutrients and used in lithium-iron-phosphorus batteries (Roy *et al.*, 2022). In rare earth elements (REEs) deposits, the REEs are often bound to P, and REEs are important for batteries, electronics and magnets (Wang & Liang, 2014; Yang *et al.*, 2017). The plant micronutrient boron is also relevant to the future of battery technologies, as boron can improve battery thermal stability (Zhang *et al.*, 2021). It is essential that important molecules like N and P are recycled for future food and energy security. The

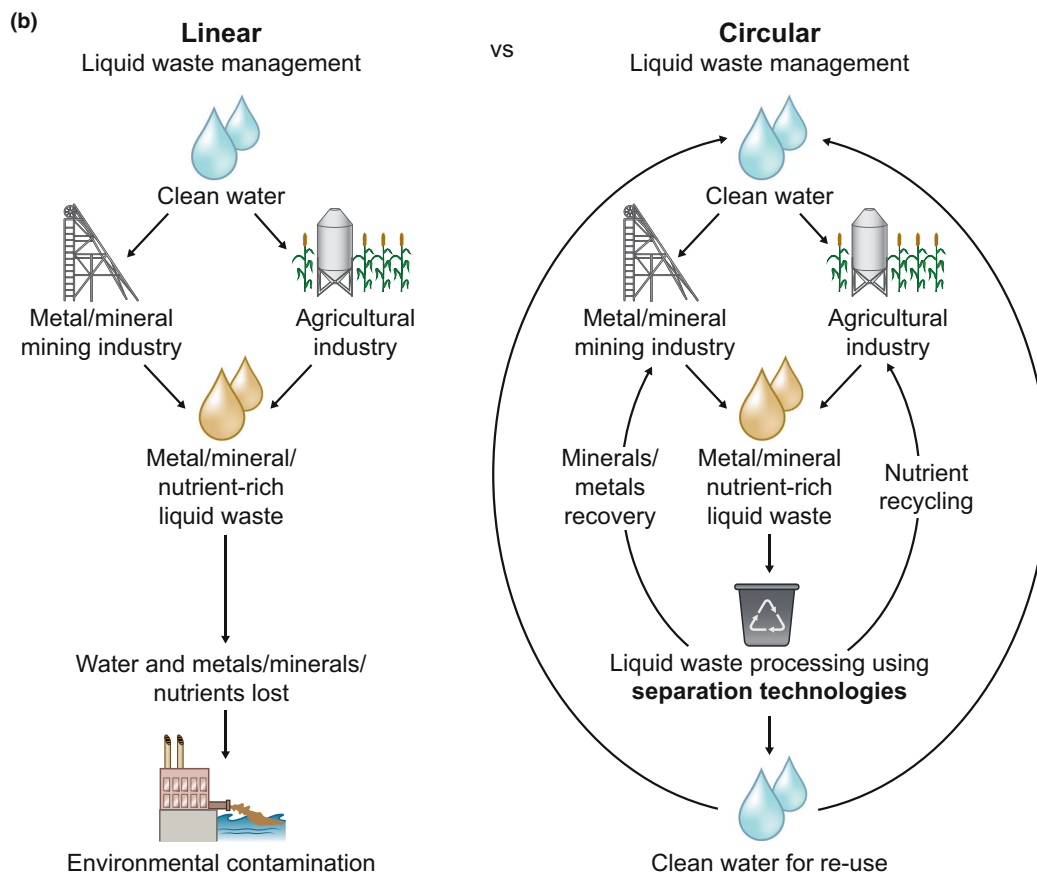
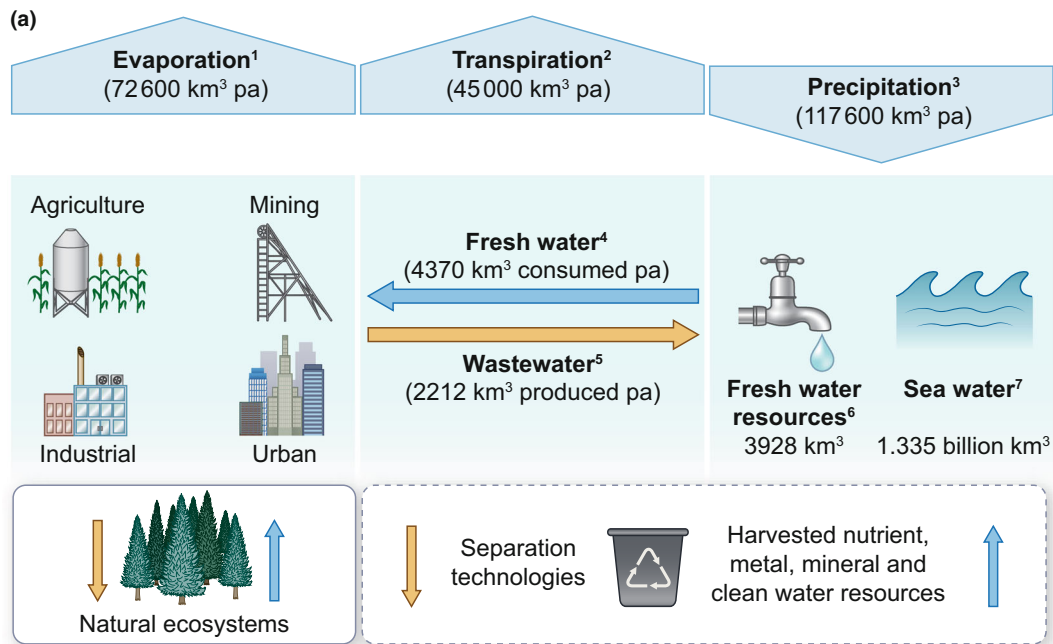
efficiency of use, reuse and recycling of these types of resources can be improved by changing resource management approaches and by implementing technologies that contribute to separating these resources from wastes, such as membrane separation technologies. Membrane separation technologies are important tools for addressing metal, mineral, nutrient and water resources security challenges and environmental challenges (T. Chen *et al.*, 2022).

The features and characteristics of plant and microbe cell membrane transport mechanisms are relevant to engineering membrane separation solutions and novel biotechnologies for improving resource management (Lee *et al.*, 2003; Bhuiyan *et al.*, 2011). Plant and microbial organisms can be used directly in phytomining systems and knowledge about the molecular mechanisms diverse organisms use to control membrane permeability can be used to advance separation technologies. For example, aquaporins are used in a range of membrane technology systems to improve membrane flux and adjust the specificity of the separation (Zhao *et al.*, 2012; Tang *et al.*, 2013; Nabeel *et al.*, 2020). There are subsets of aquaporins with high water permeability and high rejection of other molecules which have been embedded in technologies for use in water desalination (Güvensoy-Morkoyun *et al.*, 2022) and wastewater reuse applications (Tang *et al.*, 2013). There is scope for using a greater diversity of plant and microbe membrane transport mechanisms to further advance the function of separation technologies and for creating biomaterials for wastewater treatment applications (Pervez *et al.*, 2020). Tapping into these selective membrane transport protein resources could be aided by developing catalogues of diverse plant membrane transporter protein sequence, structure and permeability information (Hooper *et al.*, 2017; Newport *et al.*, 2019).

The demand for membrane systems capable of ion-selective functions is growing due to resource demand exceeding supply and

Box 2 Precision separation technology can be created by mimicking the modular nature of plant separation functions.

Modular systems offer the flexibility to use different types of separation technologies to achieve different functions at different process steps. Similarly, plants use different strategies and mechanisms in different tissues, cells and organelles to optimise separation of molecules. Embedding of selective proteins in membrane technologies is one example of a type of separation technology (Fig. 3). There has been limited testing of pairing of different combinations of separation technologies to optimise for different separation functions. Multi-modular systems where graphene filter modules, metal organic framework modules and protein-embedded membrane separation modules are brought together in a combination of organisational units can enable processing of raw materials and wastes such that different valuable components can be isolated at different steps in processing of input solutions (T. Chen *et al.*, 2022). Successful development of modular systems requires input from the target mining, manufacturing, agricultural, food or energy industries into which these systems will be applied to ensure specificity in conversion of input solutions to re-usable output products matches industry needs.



the large amounts of waste being generated around the world requiring processing (DuChanois *et al.*, 2021). There is great diversity in waste and raw material compositions which means that diversity in the function of separation technologies is required.

Separation can be achieved mechanically, biologically or chemically. There is a growing demand for the development of more sophisticated modular separation systems where consecutive separation and processing steps enable multiple resources to be

Fig. 4 Schematic representation of the role for separation technologies in recycling resources from wastewater in the context of global annual fluxes of water and management options. Each year more than 100 000 kilometres cubed (Km^3) of water (blue shaded arrows) moves between the atmosphere and Earth's crust via three processes: evaporation, transpiration through plants and precipitation (values per annum are approximate and derived from: ¹the difference between transpiration data from ²Evaristo *et al.* (2015) and precipitation data from ³Schneider *et al.* (2017), and data from ⁴Jaramillo & Destouni (2015), ^{5,6}Smol (2022) and ⁷Qadir *et al.* (2022).) (a). Healthy natural plant and microbe ecosystems play key roles in the cycles that enable fresh water resources to be replenished (Jasechko, 2018). Fresh water resources are consumed in human activities such as agriculture, mining, industrial activities and to support urban environments, and wastewater (yellow shaded arrows) is produced. A greater volume of freshwater is consumed per annum (p.a.) than the volume of fresh water resources that can be replenished by natural ecosystems. The volume and composition of the wastewater produced by human activities exceeds the capacity and capability, respectively, of the cleaning functions of natural ecosystems. Development and deployment of separation technologies that enable harvest of nutrient, metal, mineral and clean water resources from wastewater is required for future water and resource security, to prevent natural ecosystems from being further overwhelmed by wastewater and to enable transition from linear to circular liquid waste management (b) (Tarpeh & Chen, 2021). For examples of types of molecules that plants separate that may be relevant to engineering novel industrial separation technologies, see Table 1. Abbreviations: cubic kilometres (km^3), per annum (p.a.). Bin with the universal recycling symbol in the dashed box in part (a) represents where separation technologies can be used to process wastewater to harvest clean water and nutrient, metal and mineral resources to return these resources into circulation rather than releasing industrial wastewater into natural ecosystems (solid box). Droplet shapes in part (b) represent either clean water (blue shading) or wastewater (yellow shading).

retrieved from raw materials and wastes during processing (Box 2). Retrieval of nutrient, metal, mineral and clean water resources from wastes is important for building a sustainable future, supporting agricultural productivity and the capability to supply renewable energy needs and meet climate change mitigation goals.

Current commercially available membrane separation technologies can be used to harvest clean water from various wastewater sources, but the nutrient, metal and mineral resources within those sources of wastewater remain as undifferentiated waste. Reuse of those resources requires them to be relatively pure. Capture of purified sources of resources from waste requires development of precision separation technologies that are adapted to efficiently separate out target nutrient, metal and mineral resources from complex fluids whilst also generating clean water (Fig. 4). Efficient separation of these types of resources is a function that plants and microbes provide in natural ecosystems. Recreating these types of plant and microbe functions in technologies that can be used to manage waste from human activities is the challenge ahead.

V. Conclusions

Within the plant Kingdom resides a wealth of untapped selective membrane transport mechanism resources that can be retrieved and studied to inform crop improvement, phytoremediation strategies and advance industrial membrane separation capability. Sequence-structure-function information is needed to progress our understanding of the membrane protein structure and permeability relationships for mechanisms of interest, and to facilitate the identification and engineering of mechanisms relevant to advancing crop improvement and novel separation technologies. The synthetic biology revolution is offering the possibility of applying systematic design approaches to engineering the selectivity and permeability of molecular mechanisms, making it easier to resolve structure and function relationships and create mechanisms with fit-for-purpose functions (Lu *et al.*, 2018; Cravens *et al.*, 2019; Freemont, 2019).

Future research directions in this area are expected to include applying computational design and synthetic biology approaches to re-engineer the properties of candidate selective separation mechanisms to optimise the function of these mechanisms for use in industrial applications such as harvesting resources from wastes

(Xu *et al.*, 2020). Membrane protein structure and function relationship information is required for developing artificial selective membrane structures that mimic the functions of transmembrane transport mechanisms found in nature (Z. Zhang *et al.*, 2020). Functional testing and re-engineering of the selective separation processes found in nature are expected to contribute to enabling the development of precision systems for harvesting and reusing resources from liquid wastes.

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Competing interests

AM, CB and SM serve as co-Directors of Membrane Transporter Engineers Pty Ltd (ABN 75649462617).

Author contributions

ADR and SM contributed equally to this work.

ORCID

Caitlin Byrr  <https://orcid.org/0000-0001-8549-2873>
Annamaria De Rosa  <https://orcid.org/0000-0002-2610-9149>
Isobel Magrath  <https://orcid.org/0000-0002-9065-2015>
Samantha McGaughey  <https://orcid.org/0000-0001-6133-0415>

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