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Does Functionalisation Enhance CO$_2$ Uptake in Interpenetrated MOFs? An examination of the IRMOF-9 series

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The effect of pore functionalisation (-I, -OH, -OCH$_3$) on a series of topologically equivalent, interpenetrated metal-organic frameworks (MOFs) was assessed by both simulation and experiment. Counter-intuitively, a decreased affinity for CO$_2$ was observed in the functionalised materials, compared to the non-functionalised material. This result highlights the importance of considering the combined effects of network topology and chemical functionality in the design of MOFs for enhanced CO$_2$ adsorption.

Capturing CO$_2$ from major anthropogenic sources such as coal-fired power plants is an area of intense research. One of the main challenges for chemists is the development of novel energy-efficient gas separation processes. Several strategies have been proposed including physical adsorption$^1$, absorption by solvents$^2$, and gas separation membranes.$^3$-5 Indeed, recent work has shown that physical adsorption of CO$_2$ in MOFs is a promising low-energy alternative to conventional methods.$^3$ A commonly employed strategy to enhance CO$_2$ adsorption capacity at low pressures in MOFs is to incorporate moieties that have a high affinity for CO$_2$. As a result, such materials give rise to significant enthalpies of adsorption.$^2$, $^3$ In addition to tailoring the functionality of the framework, careful control over the internal pore size is important as there is a trade-off between maximising the potential energy well overlap between the pore walls at close distances, and loss of available pore volume.$^4$ For example, although framework interpenetration reduces the gravimetric gas uptake of a given material it has been shown to assist CO$_2$ uptake, highlighting the importance of tailoring both functionality and pore confinement.$^5$ In the case mentioned, the unusual and enhanced CO$_2$ adsorption shown by the framework is linked to the stepwise filling of pores and structural vacancies.

Modelling and experimental studies have shown that pore functionalisation generally improves the CO$_2$ uptake in non-interpenetrated frameworks.$^2$, $^6$-13 However, we note that the effect of functional groups on CO$_2$ capacity in interpenetrated MOFs is not well documented.$^{14}$ A recent study showed that combining both interpenetration and functionality in a MOF can enhance both H$_2$ and CO$_2$ uptake, taking advantage of their individual effects.$^{15}$ Here however, we show, for interpenetrated [Zn$_4$(BPDC)$_3$]$_x$ IRMOF derivatives (Scheme 1), that these effects do not necessarily combine to enhance CO$_2$ uptake. Accordingly, careful consideration of both the framework architecture and functionality is required for the development of suitable physisorbents for CO$_2$ capture and separation.

We previously demonstrated that by carefully controlling reaction conditions, phase-pure isostructural MOFs with interpenetrated and non-interpenetrated structures can be synthesised.$^{16}$ Here we expand this study by investigating a series of, topologically analogous, interpenetrated MOFs constructed from functionalised biphenyl links (Scheme 1) with the aim of understanding the impact of functional groups on CO$_2$ uptake.$^{16}$ Notably, our results show that irrespective of the functional group, a decrease in CO$_2$ uptake is observed when compared to that of the non-functionalised material, IRMOF-9. This
is quite surprising given that functionalisation in non-interpenetrated frameworks is generally observed to enhance CO₂ affinity. Here we use molecular modelling to shed light on how functional groups block the inaccessible regions (i.e. small pores) present in functionalised IRMOF-9 analogues (Fig. 1). This highlights that the increased potential-well overlap due to small pores is more important energetically than the presence of functional groups in enhancing CO₂ uptake, particularly at low pressure.¹⁷-¹⁹

![Scheme 1](image1)

**Scheme 1**

IRMOF-9 [Zn₄O(BPDC)₃] and its functionalised analogues (-OH, -OCH₃) were prepared as reported in previous work.¹⁶ Iodo functionalised IRMOF-9 (IRMOF-9-I) was synthesised using a slightly modified method (See ESI†). The functionalised IRMOF-9 materials are an isostructural series consisting of Zn₄O tetrahedral clusters acting as six-connecting nodes with the functional groups directed into the smaller pores that are formed by interpenetration.¹⁶ CO₂ and N₂ gas isotherms were measured for each MOF (Table 1 and Fig. 2) and Grand Canonical Monte Carlo (GCMC) simulations were employed to predict pure CO₂ isotherms and NVT simulation to predict the heat of adsorption at infinite loading (See ESI†).

**Table 1.** Framework density, pore volumes, accessible volumetric surface area and isosteric heat of adsorption at low coverage predicted from NVT simulation with blocking of the inaccessible pore regions (except for IRMOF-9-0H).

<table>
<thead>
<tr>
<th>Structure</th>
<th>(\rho_i) (g/cm³)</th>
<th>(V_{\text{free}}) (cm³/g)</th>
<th>Volumetric surface area (m²/cm³)</th>
<th>(Q_{\text{st}}) CO₂ (kJ/mol)</th>
</tr>
</thead>
<tbody>
<tr>
<td>IRMOF-9</td>
<td>0.76</td>
<td>1.18</td>
<td>1918</td>
<td>22.73</td>
</tr>
<tr>
<td>IRMOF-9-OH</td>
<td>0.705</td>
<td>0.99</td>
<td>1619(1262)¹</td>
<td>16.95</td>
</tr>
<tr>
<td>IRMOF-9-OCH₃</td>
<td>0.773</td>
<td>0.84</td>
<td>1573(1317)⁺</td>
<td>18.55</td>
</tr>
<tr>
<td>IRMOF-9-I</td>
<td>0.878</td>
<td>0.76</td>
<td>1730(1184)⁺</td>
<td>15.94</td>
</tr>
</tbody>
</table>

¹ Experimental BET surface area based on N₂ isotherms

BET analysis of the 77K N₂ isotherm for each MOF showed a reduction in surface area for the functionalised materials. This observation is in qualitative agreement with the predicted surface area based on nitrogen as a probe molecule (Table 1). Excepting IRMOF-9, typical Type I isotherms were observed for all the functionalised analogues. Due to weak inter-framework interactions in IRMOF-9 (Fig. 1), evidence of gating behaviour was observed in the N₂ as well as CO₂ isotherms at 195 K (Fig. S3 and S4, ESI†). Such gating behaviour has been found recently in a two-fold interpenetrated framework and also in a partially interpenetrated framework.⁵

In the present case, structural transformations occur under pressure for CO₂ showing multiple steps at 195 K. It is noteworthy that this is the first time such an adsorption profile has been reported for IRMOF-9. We attribute this to careful handling and activation procedures (See ESI†). Addition of functional groups to the IRMOF-9 backbone enhances the inter-framework interactions (Fig. 1) and eliminates adsorption gating behaviour. Similar behaviour has been observed in many interpenetrated MOFs with pendant groups attached to the BPDC linkers.²¹-²³ The Type I isotherms suggest that functionalised IRMOF-9 derivatives, behave more like a rigid open framework due to a lack of dynamic motion.

Next, we investigated the CO₂ gas adsorption in IRMOF-9 and its functionalised analogues at room temperature (Fig. 2). No adsorption/desorption hysteresis was observed for either IRMOF-9 or the functionalised frameworks at temperatures above the triple point. Fig. 2 shows good agreement between the simulated data and the experimental values for IRMOF-9 and the functionalised materials, when the inaccessible region is blocked in the simulation (see Fig. S11 and S12). The experimental CO₂ uptake of IRMOF-9 was measured to be 40 cm³ g⁻¹ at 100 kPa and was found to decrease drastically to 28 cm³ g⁻¹ in IRMOF-9-OCH₃ and to 16 cm³ g⁻¹ in IRMOF-9-I, respectively.

![Fig. 2](image2)

**Fig. 2** CO₂ uptake in interpenetrated IRMOF-9 and functionalised IRMOF-9 frameworks at 298 K. Simulations show that blocking the inaccessible pore in the functionalised IRMOF-9 materials provides excellent agreement to experiment. The empty symbols represent experimental data, while the solid lines represent simulated data.

This observation can be rationalised from an energetic viewpoint as the calculated adsorption enthalpy at low coverage from NVT simulations and dispersion corrected density functional theory is highest in IRMOF-9, followed by -OCH₃, -OH and -I frameworks (Table 1 and Table S7). As amino groups are widely employed for enhancing CO₂ uptake, we further predicted CO₂ uptake using GCMC simulation and found the CO₂ capacity in IRMOF-9-NH₂ is lower than that of IRMOF-9 (See Fig. S13, ESI†). We posit that the decrease in CO₂ uptake observed for the functionalised IRMOF-9 frameworks can be attributed to the blockage of small pores by the attached functional group and presence of pores with larger diameters.
larger isosteric heat of adsorption. A similar trend has been observed earlier for CO$_2$ adsorption in C$_{168}$ and N$_2$ and CO in AlPO$_4$-5 structures and CO$_2$ in an interpenetrated MOF.\textsuperscript{24,25} However, we also modelled the heat of adsorption for IRMOF-9 using virial equation methods (Fig. S17). Using this approach the initial enthalpy of adsorption is, also, approximately 22 kJ/mol and decreases slightly over the coverage range. However, the dip at low pressure is absent.

![Fig. 3 Pore size distributions based on N$_2$ isotherms showing different pore sizes in IRMOF-9, but only one distinct pore in functionalised IRMOF-9 frameworks.](image1)

Fig. 3 shows the pore size distribution obtained from the N$_2$ isotherm data of IRMOF-9 and its functionalised forms. As shown in Fig. 3 the pore sizes present in IRMOF-9 are centred around 5.8 Å and 8.0 Å (larger pores not shown in Fig. 3), however, for the functionalised analogues only one distinct pore is present centred at 11 or 12 Å. Presumably, this is because the pendant functional groups enhance the Van der Waal’s and electrostatic interactions between the frameworks creating a more uniform pore size. To examine the location of CO$_2$ in the IRMOFs, density contours of CO$_2$ obtained from GCMC simulations were prepared (Fig. S14, ESI†). In IRMOF-9, the CO$_2$ is exclusively located in the small pores created by the interpenetration of frameworks where the effective potential energy well-overlap occurs between metal clusters and phenyl linkers. However, in the functionalised IRMOF-9 forms, the CO$_2$ is mainly located in the larger pores in close proximity to the functional group, as the equivalent small cavities are inaccessible for gas molecules, even for H$_2$ (See Fig S15 ESI†). In addition, the binding energy of one CO$_2$ molecule from first principles DFT calculations displayed a similar behaviour implying that the CO$_2$ uptake correlates well with the CO$_2$-functional group interactions (Fig. S16 and Table S7, ESI†).

![Fig. 4 Isosteric heat of adsorption as a function of simulated CO$_2$ uptake using Clausius-Clapeyron equation.](image2)

Fig. 4 shows the predicted isosteric heat of adsorption from the Clausius-Clapeyron equation based on simulated adsorption isotherms at 273 K and 298 K. Consistent with CO$_2$ adsorption capacity shown in Fig. 2, the heat of adsorption decreases in the order IRMOF-9-OCH$_3$ > -OH > -I. The key result in this instance is that the increased relevance of small pores in IRMOF-9 supersedes the chemical attraction of functional groups in the derivatised analogues, as these groups are largely occluded within the interpenetrated structure. Secondly, in IRMOF-9, variations of enthalpy strength at low loading are observed. The initial decrease in isosteric heat may attributed to the multiple pore sizes present in IRMOF-9\textsuperscript{26} (Fig. 3) in which the more energetically favourable and smaller pores are occupied first followed by the less favourable sites as loading increases. As the gas loading is increased, the co-operative interaction between CO$_2$ molecules is enhanced resulting in a

Notes and references

This research is supported by the Science and Industry Endowment Fund (SIEF). CJD and CJS would like to acknowledge the Australian Research Council for funding FT100100400 and FT0991910, respectively.

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† Electronic Supplementary Information (ESI) available: Experimental and simulation details and additional results. See DOI: 10.1039/b000000x/